Two Energy Levels System: Negative Temperature, Specific Heat



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Two Energy Levels System: Negative Temperature, Specific heat Consider a system AE = 26 Consisting of N independent particles. u is the magnetic moment of each pointicle. They are either parallel or antiporallel to external field H. Let + E and - E are the energies associated with these orientations.

1E/ = MH

Total energy of isolated system E can

be given as
$$E = N \in = -N_{-} \in + N_{+} \in$$
Where $N = N_{+} - N_{-}$, $N = N_{+} + N_{-}$

$$N_{+} \text{ have energies } \in$$

$$N_{-} \text{ have energies } - \in$$

$$N_{+} = \frac{1}{2}(N + n)$$

$$N_{-} = \frac{1}{2}(N - n)$$

The total no. of ways or arrangements SZ(n) corresponding to a net magnetic moment M = nM and energy E = nEis $SZ(n) = \frac{IN}{IN-IN+}$

So the no. of possible microstates corresponding to macroscopic conditions is r(n).

Equal volumes of phone space are associated with each independent arrangement of the magnetic moments. If this volume is unity then probability of occurrence of a particular macrostate is M(n) = N(n).

We know that $\ln(1\pm\eta_N) = \pm \frac{n}{N} - \frac{n^2}{2N^2} \pm \frac{n}{N}$ 1: LnW =- [=N(1-1/N)ln=[1-1/N)+=N(1+1/N). In = (1+ 1/N) $- N - \left[ln \frac{1}{2} + \left(-\frac{N}{N} - \frac{N^2}{2N^2} \right) \right]$ $-N_{+}[l_{N} \frac{1}{2} + (\frac{N}{N} - \frac{N^{2}}{2N^{2}})]$ $= -N \ln \frac{1}{2} - \frac{h^2}{2N}$ Probability w(M) of a Net Moment M In zero mognetic field each For any specific arrangement, the

porticle has probability of for that orientation. .. probability that a specific arrangement of moments to occur 14 (=) N. so the probability w(M) of a net moment M $\omega(M) = \left(\frac{1}{2}\right)^N \ln(n)$ $= (\frac{1}{2})^{N} \exp \left\{ \ln W \right\}$

$$= (\frac{1}{2})^{N} \exp (\ln w \ln x)$$

$$= (\frac{1}{2})^{N} \left[(\frac{1}{2})^{-N} \exp (-\frac{h^{2}}{2N}) \right]$$

$$= \exp (-\frac{h^{2}}{2N})$$

Magnetization has Gaussian distribution about the value n=0.

In the absence of external magnetic field, the average value of magnetization is geno.

(gassian distribution

$$= -K \left[N - ln \left(\frac{N-1}{N} \right) + N + ln \left(\frac{N+1}{N} \right) \right]$$

$$= -K \left[N ln \frac{1}{2} + \frac{n^2}{2N} \right]$$

$$= -N K \left[-ln 2 + \frac{(E/E)^2}{2N^2} \right]$$

Plot of Six with respect to EN is shown in figure. The slope $\frac{\partial S}{\partial E}$ gives the T=0.

Sign of temperature T. T=0. T=0. $\frac{1}{T} = \left(\frac{\partial S}{\partial E}\right)^{V,N}$ $=\frac{1}{100}\left(\frac{9c}{9u}\right)^{1/2}=\frac{1}{100}\frac{1}{100}\left(\frac{10-10}{100}\right)$

Negative Temperature

At conventional zero temperature, all particles are in ground state i.e in -E and $N = N_{+} - N_{-} = 0 - N = -N < 0$ It gives $T = 0_{+}$ and S = 0

Complete order at E = -NE As temperature increases or energy is given to the system, N+ grows giving T>0 till n=0 where s= NKln2 maximum disorder and T = 00+ If more energy is given to the system, population in upper level increases, i-e N+>N- n>0 we get dicreense in & more order normal behaviour. This temperature i.e. negative temperature T_ Corresponds to higher energy than positive temperature T. Further increasing energy given to the system, $N = N_+ - N_- = N - 0 = N > 0$ all particles in the upper state and S = 0 complete order and T = 0.

In the graph, at one side negative temperatures varies from -00 to-0 and positive temperatures varies from +0 to +00 is represented where +00 and -00 coincide with eachother.

Specific heat
$$\frac{N+n}{N-n} = \frac{N+}{N-}, \quad \Delta E = 2E$$

$$\frac{N+}{N-} = \exp\left(\frac{-2E}{KT}\right) = \exp\left(\frac{-\Delta E}{KT}\right)$$

$$\frac{N-}{N} = \frac{\exp\left(\frac{E}{KT}\right) + \exp\left(\frac{-E}{KT}\right)}{\exp\left(\frac{E}{KT}\right) + \exp\left(\frac{-E}{KT}\right)} = \frac{1}{1 + \exp\left(\frac{\Delta E}{KT}\right)}$$

$$\frac{N+}{N} = \frac{\exp\left(\frac{E}{KT}\right) + \exp\left(\frac{-E}{KT}\right)}{\exp\left(\frac{E}{KT}\right) + \exp\left(\frac{-E}{KT}\right)} = \frac{1}{1 + \exp\left(\frac{\Delta E}{KT}\right)}$$

$$energy \quad E = nE = -\left(N-\frac{N+}{N+}\right)E$$

energy
$$E = nE = -(N_--N_+)E$$

= -NEtanh($\frac{E}{RT}$)

Specific heat $\left[e_{\times}b\left(\frac{\epsilon}{\kappa_{T}}\right)+e_{\times}b\left(\frac{-\epsilon}{\kappa_{T}}\right)\right]^{2}$ NK $\left(\frac{\Delta E}{KT}\right)^2$ $exp\left(\frac{\Delta E}{KT}\right)$ $\int 1 + exp\left(\frac{\Delta E}{KT}\right)^2$ Schottky anomaly Specific heat show a peak. Gic heat shows a peak body how a gop DE in energy states.

References:

- Statistical Mechanics by R. K. Pathria
- Statistical Mechanics by B. K. Agarwal and M. Eisner
- An Introductory Course of Statistical Mechanics by P. B. Pal
- Elementary Statistical Physics by C. Kittel
- Fundamentals of Statistical and Thermal Physics by F. Reif
- Statistical and Thermal Physics by R. S. Gambhir and S. Lokanathan

Thank You

For any questions/doubts/suggestions and submission of assignments

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